A SHORT INTRODUCTION TO TWO-PHASE FLOWS Industrial occurrence and flow regimes

Hervé Lemonnier DM2S/STFM/LIEFT, CEA/Grenoble, 38054 Grenoble Cedex 9 Tél. 04 38 78 45 40 herve.lemonnier@cea.fr, herve.lemonnier.sci.free.fr/TPF/TPF.htm

ECP, 2011-2012

CLASSES CONTENTS (1/2)

- Introduction: CEA/Grenoble, scientific information
- Two-phase flow systems in industry and nature
- Flow regime
- Measuring techniques, composition (α)
- Simple models for void fraction prediction



CLASSES CONTENTS (2/2)

- Balance equations
- 1D models, pipe flow
- Pressure drop and friction
- Heat transfer mechanisms in boiling
- Condensation of pure vapor
- Critical flow phenomenon

Recommended textbook: Delhaye (2008)



THERMAL-HYDRAULICS

- $\bullet~$ Study of simultaneous flow and heat transfer, in French, thermohydraulique
- Phase: state of matter characterized by definite thermodynamic properties
- Two-phase: mixture of two phases (diphasique)
- Examples: air and water, oil and water (connate), water and steam, oil and natural gas (multiphase), *polyphasique*).

CEA/GRENOBLE RESEARCH CENTER

- CEA: Commissariat à l'Energie Atomique (15000 p)
- CEA/Grenoble: originates in 1956, founded by Louis Néel (4000p/2300 CEA)
- Heat transfer laboratories founded by Henri Mondin
- Nuclear energy directorate (5000 p, DEN)
- Department of nuclear technology (400 p, Cadarache, Grenoble, DTN)
- Department of reactors studies (400 p, Cadarache, Grenoble, DER)
- Labs of simulation in thermal-hydraulics (SSTH)
- Labs of experimental studies in thermal-hydraulics (SE2T)
- Thesis advising capabilities and referenced research groups for several Masters.



LABS OF SIMULATION AND EXPERIMENTS IN THERMAL-HYDRAULICS

- Codes (*logiciels*) for safety studies, CATHARE.
- 3D Codes for two-phase boiling flows (Neptune).
- LES of single-phase flow and heat transfer (TRIO-U).
- Dedicated studies : safety and optimization of NR of various generations II, III and IV, ship propulsion, cryogenic rocket engines.
- Analytic studies on boiling flows and critical heat flux (DEBORA)
- Thermal-hydraulic qualification of fuel bundles (OMEGA)
- Instrumentation development for single-phase and two-phase flows:

Can only be modeled a quantity which can be measured

 $Applications \rightarrow models \ and \ codes \rightarrow experimental \ validation \rightarrow instrumentation.$

SCIENTIFIC KNOWLEDGE AND INFORMATION

- How to solve a technical/scientic issue?
- Textbooks, books, journal papers: Library?.
- Scientific Societies: journals editing, conference organizations (proceedings, *actes*).
 - La Société française de l'énergie nucléaire
 - La Société hydrotechnique de France
 - La Société française de thermique
 - American nuclear society, thermal-hydraulics division (NURETH)
- Do you speak English? ...



TWO-PHASE SYSTEMS IN INDUSTRY AND NATURE (1/4)

- Nuclear engineering: sizing, safety, decontamination (cleaning up)
 - Loss of coolant accidents (LOCA-APRP).
 - Severe accidents w/o vessel retention.
 - Decontamination by using foam.
 - Nuclear waste reprocessing.
- Oil engineering, hot issue: two-phase production
 - Transport.
 - Pumping.
 - Metering.
 - Oil refining (Chem. Engng).
- Oil engineering : safety
 - Safety of installations.
 - LPG storage tanks and fire (BLEVE).

TWO-PHASE SYSTEMS IN INDUSTRY AND NATURE (2/4)

- Chemical engineering
 - Wastewater treatment (interfacial area and residence time).
 - Gas-liquid reactors (falling film, trickle bed, air-lift).
 - Mixing and separation.
 - Safety: homogeneous thermal runaway.
- Automotive industry
 - Diesel fuel atomization.
 - Combustion in diesel engines.
 - Cavitation damage : power steering, fuel nozzles.
- Heat exchangers
 - Condensers and evaporator/steam generator.
 - Boilers (critical heat flux, CHF), heaters.

TWO-PHASE SYSTEMS IN INDUSTRY AND NATURE (3/4)

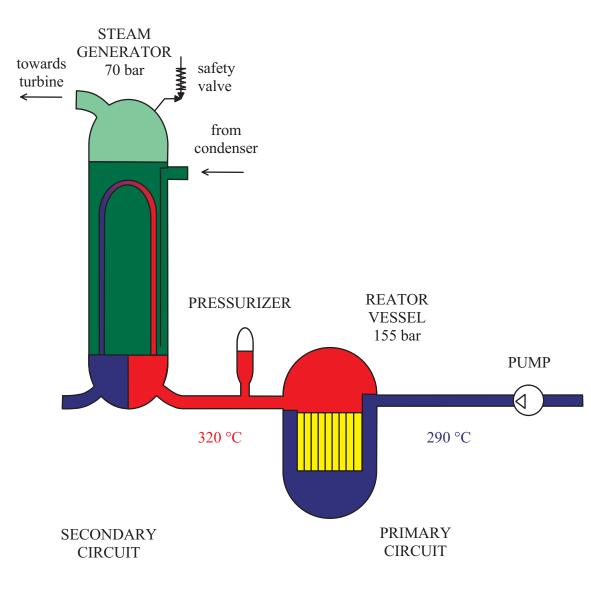
- Hydroelectricity and water distribution
 - Water resources management: transients of pipings.
 - Priming of siphons.
- Space industry
 - Cryogenic fuel storage (Vinci).
 - Thermal control of rocket engines (combustion chamber and nozzle).
 - Water hammer and pressure surges.
 - Cavitation in turbo-pumps. Instability (lateral loading) and damage.



TWO-PHASE SYSTEMS IN INDUSTRY AND NATURE (4/4)

- Meteorology
 - Storm formation, rain/hail, lightning.
 - Ocean and atmosphere exchanges, aerosols formation.
- Volcanology
 - Critical flow of lava in wells.
 - Steam explosion.
 - Nuées ardentes (Vesuvius, protection of Naples suburbs).
- Nivology
 - Avalanches.
 - Snow maturation (three-phase / 2-component).

NUCLEAR REACTORS, WATER COOLED



• Sizing :

SG: heat transfer and pressure drop.SGTR: critical flow at the safety valve.FSI: mechanical loading and

vibrations.

- Safety : LOCA, fuel cladding temperature, reference scenario
- Decontamination : vessel, SG, minimizings wastes: foam

NUCLEAR FUEL



- Fuel pellet.
- Rod ≈ 10 mm in diameter (first confinement barrier).
- Fuel assembly 17×17 .
- Control rods.
- Length ≈ 4 meters.
- Reactor core ≈ 4 m in diameter.
- Heat transfer: forced convection (7 m/s.)
- Thermal power $3000 \div 5000$ MW.

STEAM GENERATOR



- Tube-type, separates primary and secondary circuits (second confining barrier).
- ≈ 5000 tubes, diameter 50 mm, height 10 m.
- Pressure: 155-70 bar.
- 3 or 4 SG and flow loops.
- Inverted U-Tubes.
- Secondary : two-phase flow.
- Issues : heat transfer and vibration damage.



THE N4 PWR: SOME FIGURES

- Primary side pressure: 155 bar, $T_{\rm sat} \approx 355^{\circ}$ C. Thermal power: 4250 MW.
 - Mass flow rate: 4928,6 kg/s per SG (4)
 - Core inlet temperature: 292,2°C
 - Core outlet temperature: 329,6°C
- Secondary side, SG vapor pressure : 72,3 bar
 - Vapor temperature: 288 $^\circ\,$ C
 - Feed water temperature: 229,5°C
 - Mass flow rate: 601,91 kg/s per SG (4)
- Assess the thermal balance of reactor and SG

Source: National Institute of Standards and Technology (NIST) (http://webbook.nist.gov/chemistry/fluid/)



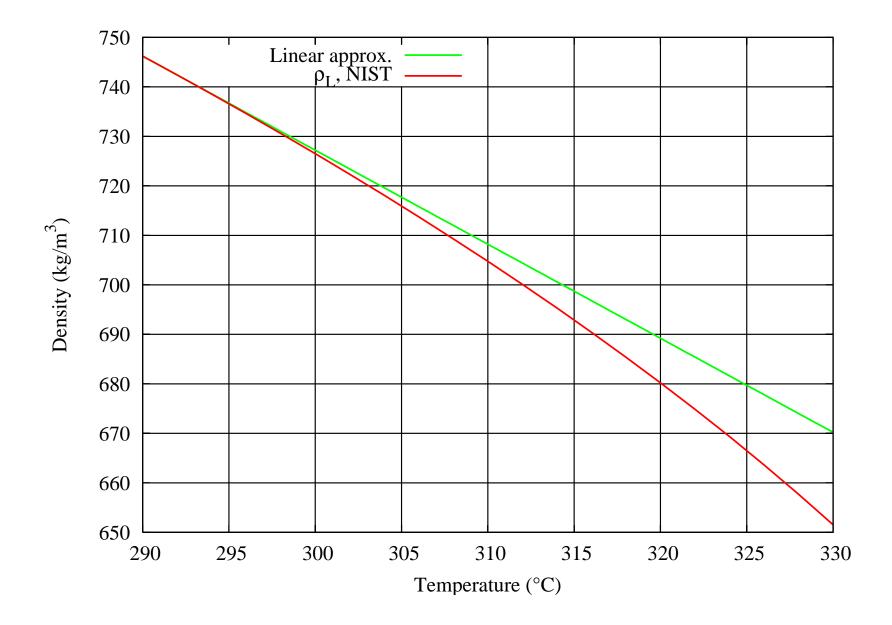
SOME BAD NEWS...

The following (low pressure) statements are *rather* wrong:

- The mass balance reads, $Q_1 = Q_2$, since water is incompressible, at least weakly it is dilatable.
- The enthalpy is, $h = C_P T$.
- For a liquid, $C_P \approx C_V$, or $h \approx u$.
- Steam is a perfect gas.



WATER DENSITY AT 155 BAR



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MASS BALANCE OF THE PRIMARY CIRCUIT

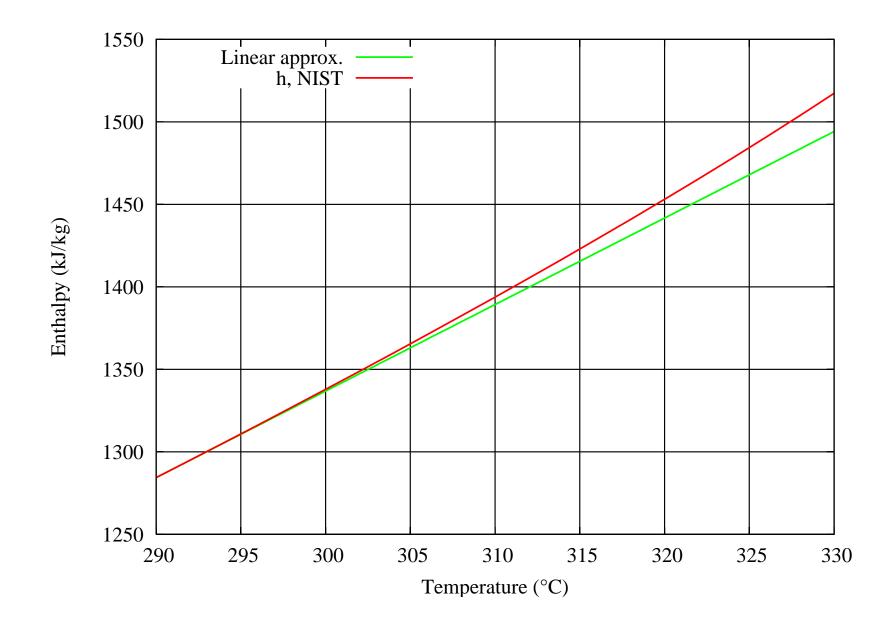
- Mass flow rate per loop (CL), $M_L \approx 5023 \text{ kg/s}$
- Inlet density: $\rho_{L1}(292^{\circ}\text{C}, 155 \text{ bar}) = 742, 41 \text{ kg/s}.$
- Outlet density : $\rho_{L2}(330^{\circ}\text{C}, 155 \text{ bar}) = 651, 55 \text{ kg/s}.$

$$Q_1 = \frac{M_L}{\rho_1} = 6,77 \text{m}^3/\text{s}, \quad Q_2 = \frac{M_L}{\rho_2} = 7,71 \text{m}^3/\text{s}$$

- Volumetric flow rates differ by 13%.
- The volume of the primary circuit is 400 m^3 ...



WATER ENTHALPY AT 155 BAR



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PRIMARY SIDE HEAT BALANCE

- Mass flow rate per loop (CL), $M_L \approx 5023$ kg/s.
- Inlet enthalpy: $h_{L1}(292^{\circ}\text{C}, 155\text{bar}) = 1295 \text{ kJ/kg}.$
- Outlet enthalpy: $h_{L2}(330^{\circ}\text{C}, 155 \text{ bar}) = 1517 \text{ kJ/kg}.$

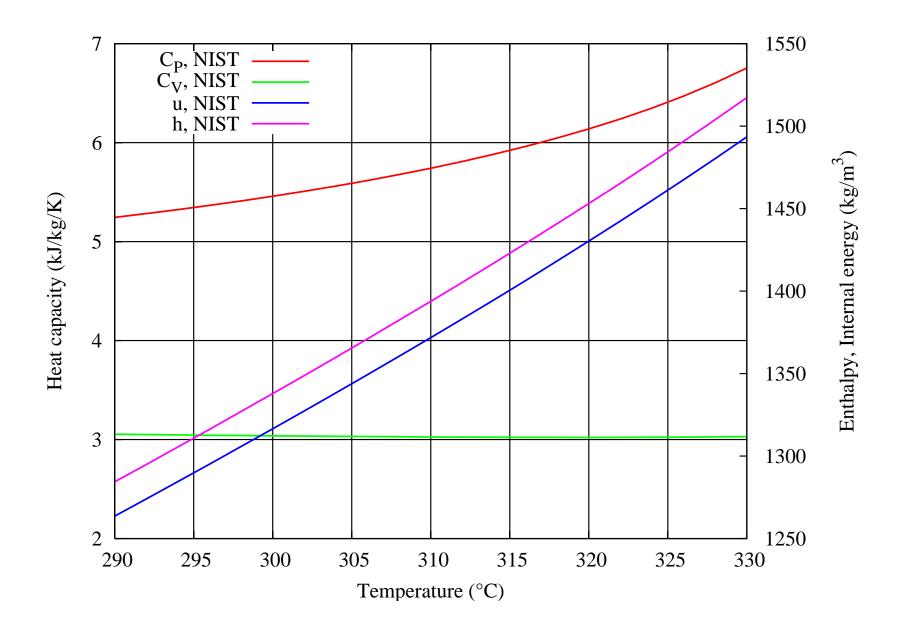
 $P = M_L \Delta h \approx 5023 \times 222 \, 10^3 = 1115 \text{ MW}$

- 4-loop reactor power: 4460 MW.
- Linear approximation: $h = C_P T$, $C_P(292^{\circ}\text{C}, 155\text{bar}) = 5,2827 \text{ kJ/kg/K}$.

 $P = M_L C_P \Delta T \approx 5023 \times 201 \, 10^3 = 1008 \text{ MW}$

- Power differs by 10%.
- Temperature drift: 31°C/hour.

WATER ENTHALPY & INTERNAL ENERGY AT 155 BAR



PHASE CHANGE & METASTABLE STATES

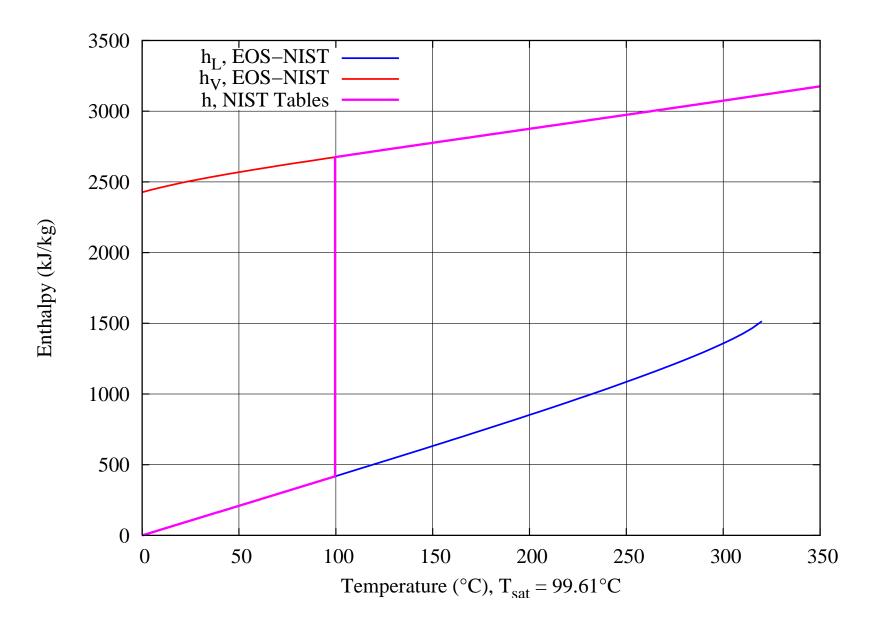
• Phase rule: thermodynamic equilibrium of steam and vapor, temperature and pressure are linked, saturation states

$$v = n + 2 - \varphi = 1$$
, $p = p_{sat}(T)$, or $T = T_{sat}(p)$

- Phase coexistence pressure, resp. temperature
- Liquid only is thermodynamically stable above $T_{sat}(p)$.
- Vapor only is thermodynamically stable below $T_{\text{sat}}(p)$.
- Water must boil and comply with the second principle of thermodynamics
- Within the metastability T range, the fluid can be either two-phase or single-phase.
- Metastable states usually not in tables, EOS is needed.



WATER ENTHALPY AT 1 BAR

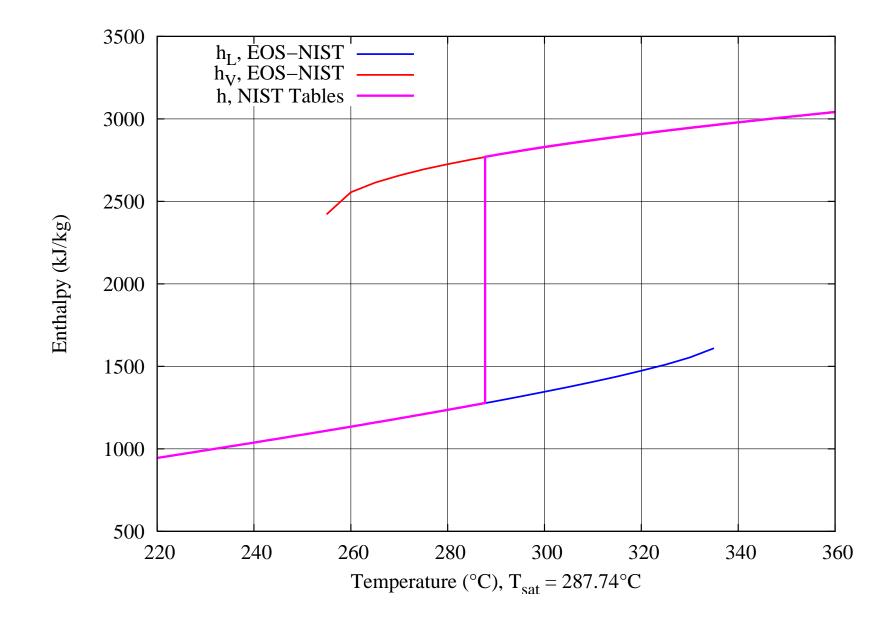


Maximum liquid superheat, metastability limit >230 K

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WATER ENTHALPY AT 72 BAR



Industrial occurrence and flow regimes

SECONDARY SIDE HEAT BALANCE

- Mass flow rate per SG, $M_L \approx 602 \text{ kg/s}$
- Inlet enthalpy: $h_{L1}(230^{\circ}\text{C}, 72 \text{ bar}) = 991.1 \text{ kJ/kg}.$
- Outlet enthalpy: $h_{V2}(288^{\circ}\text{C}, 72 \text{ bar}) = 2771 \text{ kJ/kg}.$

 $P = M_L \Delta h \approx 602 \times 1780 \, 10^3 = 1071 \text{ MW}$

• To be compared to 1115 MW on the primary side.



STEAM IS A PERFECT GAS

- Steam density at 100°C, 1 bar, $\rho_V = 0,5897 \text{ kg/m}^3$.
- Perfect gas approximation: pV = RT, R = 8.316 J/mol/K, M = 18 g/mol.

$$V = \frac{RT}{p} = 3.103 \, 10^{-2} \, \mathrm{m}^3, \quad \rho = \frac{M}{V} = 0.5801 \, \mathrm{kg/m^3}.$$

- Steam at 288°C, 72 bar, $\rho_V = 37.64 \text{ kg/m}^3$.
- Perfect gas : pV = RT, R = 8.316 J/mol/K, M = 18 g/mol.

$$V = \frac{RT}{p} = 6.481 \, 10^{-4} \, \mathrm{m}^3, \quad \rho = \frac{M}{V} = 27.77 \, \mathrm{kg/m^3}$$

• Perfect gas approximation under-estimate by 26%.

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TWO-PHASE FLOW VARIABLES

- Phase presence function
- Space averaging operators
- Instantaneous flow rates
- Time averaging operators
- Some mathematical properties
- Averaged flow rate and superficial velocity



PHASE PRESENCE FUNCTION

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(**r**)

$$X_{k}(\mathbf{r}, t)$$

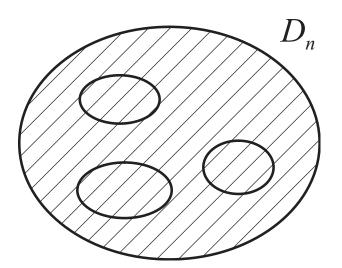
$$X_{k}(\mathbf{r}, t) = \begin{cases} 1 & \text{if } x \in \text{phase } k \\ 0 & \text{if } x \notin \text{phase } k \end{cases}$$

Measurable variable,

- Resistive probe (electrical impedance)
- Optical probe (refraction index)
- Thermal anemometry (heat transfer)

Subscripts : k = 1, 2, k = L, G, k = f, v etc.

SPACE AVERAGING (1/3)

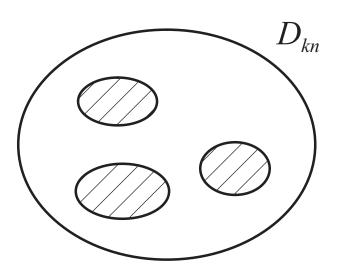


Space averaging operator (plain space average)

$$\langle f \rangle_n \triangleq \frac{1}{D_n} \int_{D_n} f \, \mathrm{d}D_n$$

- n=1, line (chord in a pipe)
- n=2, surface (cross section)
- n=3, volume (some length of a pipe)

SPACE AVERAGING (2/3)



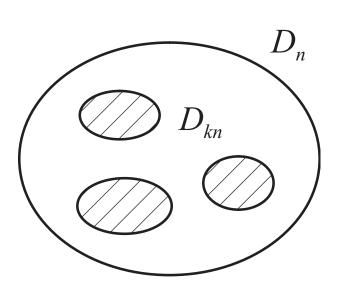
Phase presence conditional average,

$$\langle f_k \rangle_n \triangleq \frac{1}{D_{kn}} \int_{D_{kn}} f_k \, \mathrm{d}D_{kn}$$

- n=1, line
- n=2, surface (shown here)
- n=3, volume

SPACE AVERAGING (3/3)

Instantaneous phase fraction



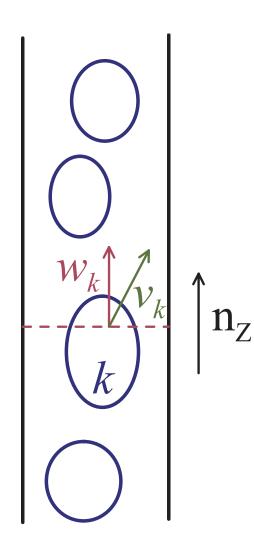
$$R_{kn}(t) \triangleq \langle X_k(\mathbf{r}, t) \rangle_n = \frac{D_{kn}}{D_n}$$

• n=1, line fraction, $\frac{L_k}{L_1 + L_2}$
• n=2, surface fraction, $\frac{A_k}{A_1 + A_2}$
• n=3, volume fraction, $\frac{V_k}{V_1 + V_2}$

Identity (proof left as an exercise)

$$R_{kn} < f >_{kn} = \langle X_k f \rangle_n$$

FLOW RATE AND MASS FLOW RATE



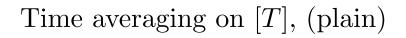
• Instantaneous flow rate (m³/s), $w_k = \mathbf{v}_k \cdot \mathbf{n}_z$

$$Q_k(t) \triangleq \int_{A_k} w_k \, \mathrm{d}A_k = A_k < w_k >_2$$

• Instantaneous mass flow rate (kg/s)

$$M_k(t) \triangleq \int_{A_k} \rho_k w_k \, \mathrm{d}A_k = A_k < \rho_k w_k >_2$$

TIME AVERAGING



$$\overline{f}(t) \triangleq \frac{1}{T} \int_{t-T/2}^{t+T/2} f(\tau) \,\mathrm{d}\tau$$

Conditional time averaging, $[T_k]$

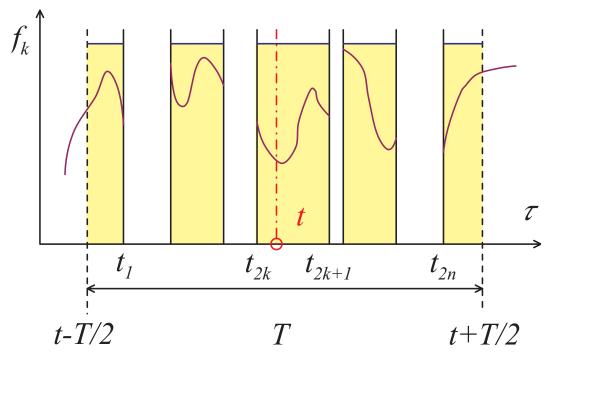
$$\overline{f}_k^X(t) \triangleq \frac{1}{T_k} \int_{[T_k]} f_k(\tau) \,\mathrm{d}\tau$$

Local time fraction, *void fraction* for gas/vapor

$$\alpha_k(\mathbf{r},t) \triangleq \frac{T_k}{T} = \overline{X_k(\mathbf{r},t)}$$

Identity, derives from the definitions,

$$\alpha_k \overline{f}^X = \overline{X_k f}$$



 $[T_k]$

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COMMUTATIVITY OF AVERAGING OPERATORS

$$\overline{R_{kn} < f_k >_n} = \langle \alpha_k \overline{f_k}^X \rangle_n$$

Proof, from definitions:

$$\overline{R_{kn}} < f_k >_n = \frac{1}{T} \int_{[T]} \left\{ \frac{R_{kn}}{D_{kn}} \int_{D_{kn}(t)} f_k \, \mathrm{d}D_{kn} \right\} \mathrm{d}t$$
$$\frac{1}{T} \int_{[T]} \mathrm{d}t \frac{1}{D_n} \int_{D_n} X_k f_k \, \mathrm{d}D_n = \frac{1}{D_n} \int_{D_n} \mathrm{d}D_n \frac{1}{T} \int_{[T]} X_k f_k \, \mathrm{d}t$$
$$\frac{1}{D_n} \int_{D_n} \left\{ \frac{\alpha_k(\mathbf{r})}{T_k} \int_{[T_k]} f_k \, \mathrm{d}t \right\} \mathrm{d}D_n = \langle \alpha_k \overline{f_k}^X \rangle_n$$

Significant example: mean void fraction, $f_k = 1$, salami theorem...

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MEAN FLOW RATES

• Mean volume flow rate,

$$\overline{Q_k} = A\overline{R_{k2}} < w_k >_2 = A \not\leqslant \alpha_k \overline{w}_k^X \not\geqslant _2$$

• Mean mass flow rate,

$$\overline{M_k} = A\overline{R_{k2}} < \rho_k w_k >_2 = A \not\leqslant \alpha_k \overline{\rho_k w_k}^X \not\geqslant 2$$



SUPERFICIAL VELOCITY

• Mean volumetric flux,

$$j_k \triangleq \overline{X_k w_k} \equiv \alpha_k \overline{w}_k^X$$

• Mean mass flux,

$$g_k \triangleq \overline{X_k \rho_k w_k} \equiv \alpha_k \overline{\rho_k w}_k^X$$

• Superficial velocity (vitesse débitante),

$$J_k = \langle j_k \rangle_2 = \langle \alpha_k \overline{w}_k^X \rangle_2 = \frac{\overline{Q_k}}{A}$$

• Mixture superficial velocity,

$$J = J_1 + J_2 = \frac{\overline{Q_1} + \overline{Q_2}}{A}$$



QUALITY

• Mean mass flux,

$$G_k \triangleq \langle g_k \rangle_2 = \langle \alpha_k \overline{\rho_k w}_k^X \rangle_2 = \frac{\overline{M_k}}{A}$$

• Mixture mean mass flux,

$$G = G_1 + G_2 = \frac{\overline{M_1} + \overline{M_2}}{A}$$

• Quality, (*titre massique*),

$$x_k = \frac{M_k}{M}, \quad M = M_1 + M_2$$

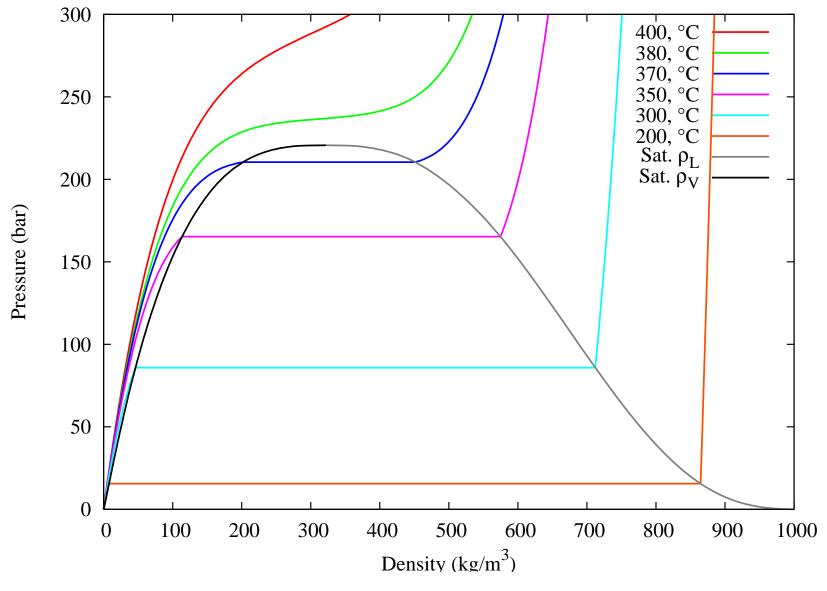
• Volume quality,

$$\beta_k = \frac{Q_k}{Q}, \quad Q = Q_1 + Q_2$$

• Equilibrium (steam) quality...

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WATER THERMODYNAMIC DIAGRAM (ρ, p)

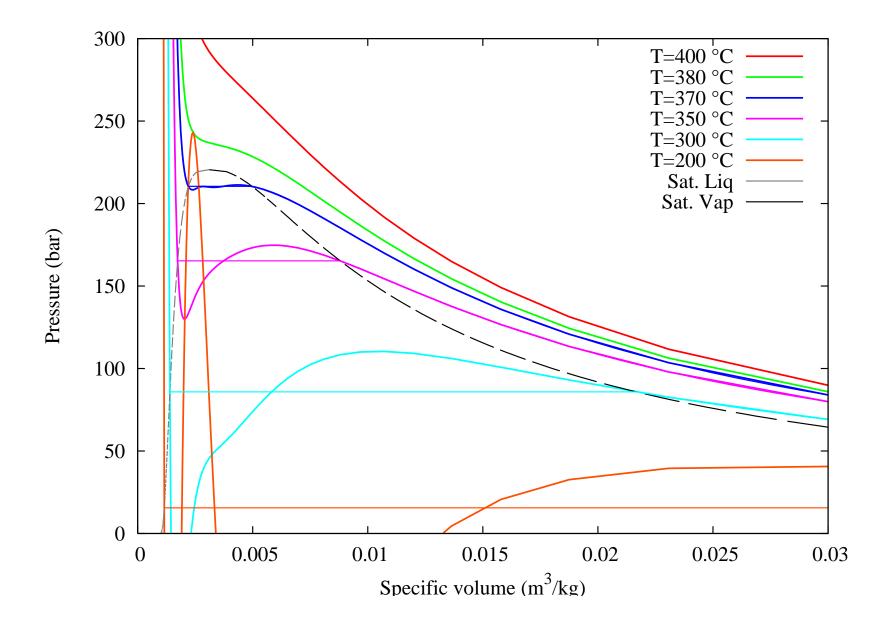


Critical pressure and temperature : ≈ 221 bar, 373,9°C.

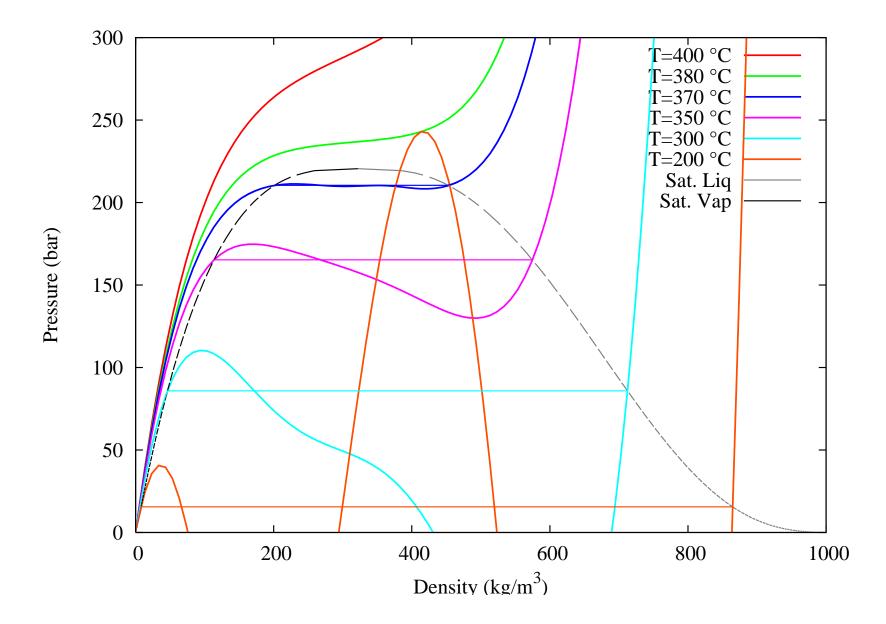
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STEAM TABLES AND EOS (v, p)



STEAM TABLES AND EOS (ρ, p)



EQUILIBRIUM QUALITY

• The thermodynamic equilibrium assumption,

 $T_L = T_V = T_{\text{sat}}(p), \quad h_k(T_k, p) = h_k(T_{\text{sat}}(p), p) \triangleq h_{k\text{sat}}(p)$

- 1D model assumption, flat profiles $(h_k(\mathbf{r}) \neq \langle k_k \rangle_2)$,
- Energy balance, q, uniform heat flux distribution

$$P = \pi q D z = M[h(z) - h_1] = M[(x_{eq} h_{V \text{sat}} + (1 - x_{eq}) h_{L \text{sat}}) - h_{L1})]$$
$$x_{eq} = \frac{h_{L1} - h_{L \text{sat}}}{h_{lv}} + \frac{\pi q D z}{M h_{lv}}$$

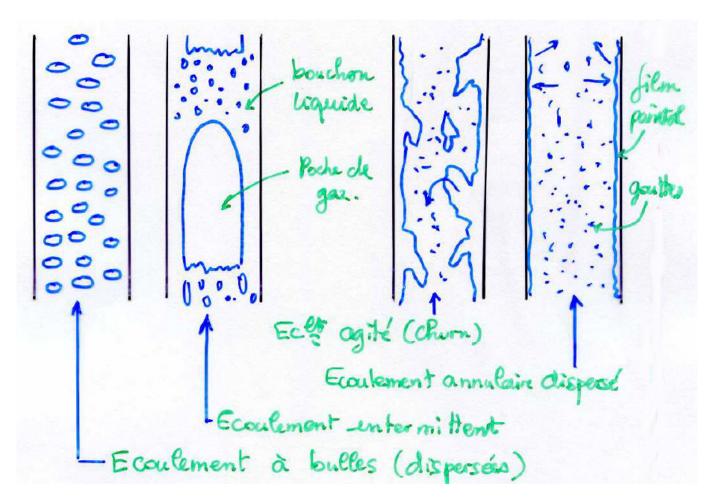
- Phase change enthalpy : $h_{lv} \triangleq h_{Vsat} h_{Lsat}$
- Equilibrium quality varies linearly with position.
- h is the mean enthalpy (energy balance), non dimensional enthalpy,

$$x_{eq} = \frac{h - h_{Lsat}}{h_{lv}}$$

FLOW REGIMES

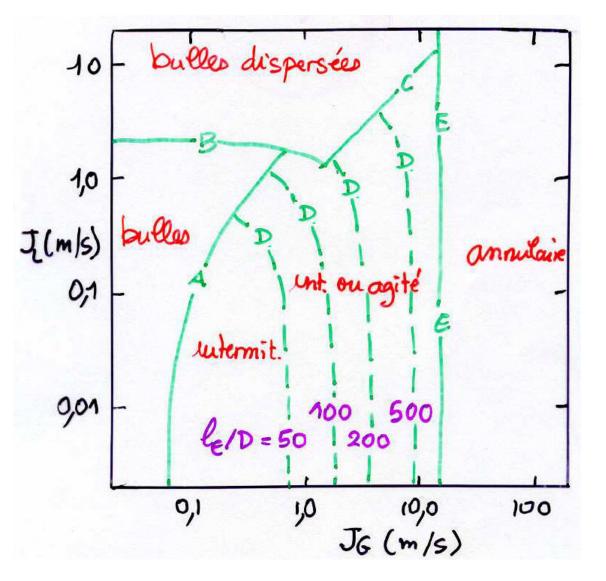
- Topological phase organisation in flows,
 - Bubbles, *bulles*
 - Plugs and slugs, poches et bouchons
 - Liquid films, drops and droplets
 - No sharp transitions
- Modeling is the motivation for identification of flow regimes
 - Single-phase flows: laminar-turbulent (NS or RANS)
 - Two-phase flows: structures of interfaces \rightarrow model.
 - Shortcomings: fully developed flows, fuzzy transitions, hydrodynamic singularities.
- Control variables: flow rates, slope, direction, diameter, transport properties, inlet conditions *etc*.
- Examples: vertical and horizontal co-current flows.

VERTICAL ASCENDING FLOWS



Flow regime transitions:

- Experiments, empirical, flow rates, momentum fluxes.
- Transition modeling, mechanisms, (Dukler & Taitel, 1986).



Flow regimes:

- A-B: Bubbly (*bulles*)
- A-D:Intermittent (poches, bouchons)
- D-E: Churn (agité)
- E: Annular
- B-C: Dispersed bubbles

Vertical ascending air-water flow, D = 50 mm,

$$P = 1$$
 bar.



• Bubbly flow and intermittent (A): Bubble coalescence, zig-zag motion of bubbles.

$$J_L = \frac{1-\alpha}{\alpha} J_G - (1-\alpha)^{\frac{3}{2}} U_{0\infty}, \quad \alpha_T = 0,25, \quad U_{0\infty} = 1,53 \left(\frac{g(\rho_L - \rho_G)\sigma}{\rho_L^2}\right)^{\frac{1}{4}}$$

• Dispersed bubbles and bubbly : turbulent break up, small bubbles, rectilinear path (B), dense packing (A) with $\alpha_T = 0,52$ (D).

$$\frac{2[\rho_L/(\rho_L - \rho_G)g]^{0.5}\nu_L^{0.08}}{(\sigma/\rho_L)^{0.10}D^{0.48}}J^{1.12} \ge 3,0$$

• Intermittent and churn (D): churn flow \equiv development of slug flow.

$$\frac{L}{D} = 42, 6\left(\frac{J}{\sqrt{gD}} + 0, 29\right)$$

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• Annular (E): all liquid entrained by the gas, force balance,

$$\frac{J\rho_G^{\frac{1}{2}}}{[\sigma g(\rho_L - \rho_G)]^{\frac{1}{2}}} = 3, 1$$

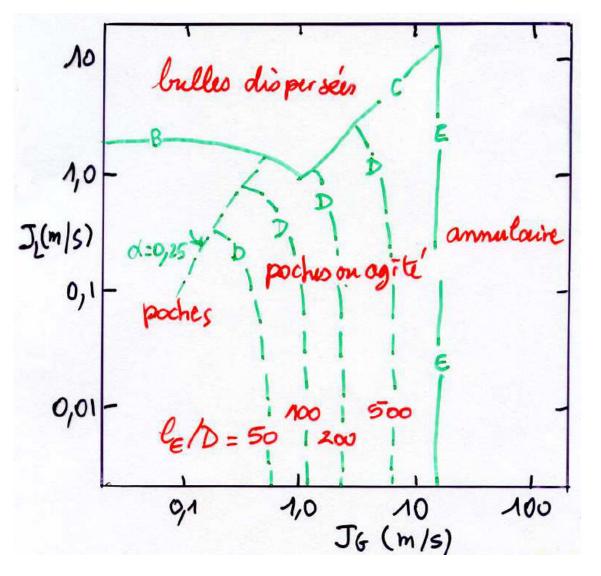
See also flooding correlations & Ku.

- Small diameter pipes: bubbles (zig-zag) and occasionally Taylor bubbles.
 - Taylor bubbles relative velocity: $U_T = 0,35\sqrt{gD}$
 - individual bubbles: $U_B = U_{0 \inf} (1 \alpha)^{\frac{1}{2}}$
 - In small diameter pipes, Taylor bubbles are slower than individual bubbles to coalescence towards slugs.

$$\left(\frac{\rho_L^2 dD^2}{(\rho_L - \rho_G)\sigma}\right)^{\frac{1}{4}} \leqslant 3,78$$

– Bubbly flow no longer exists.

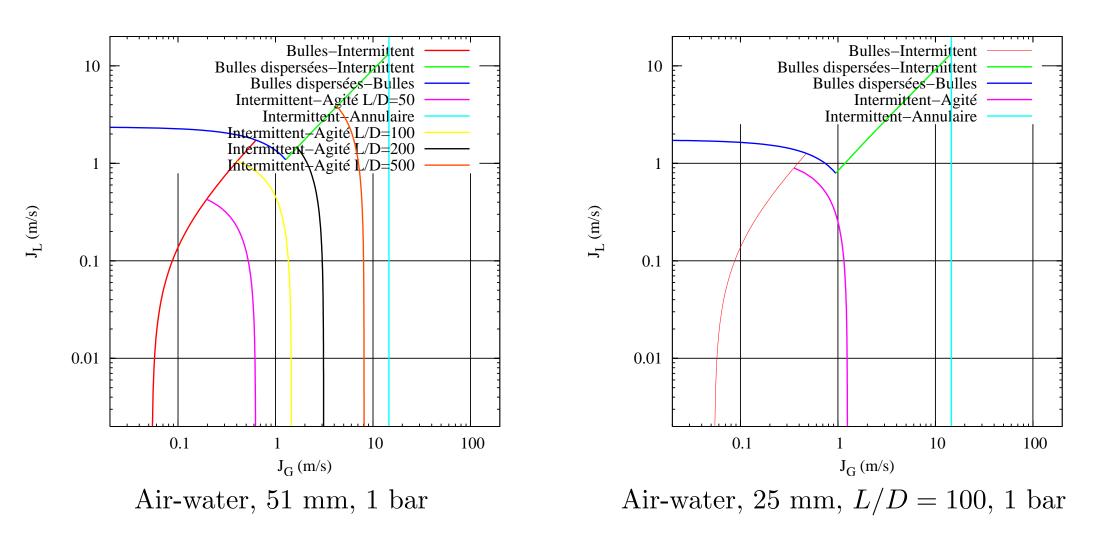




Small diameter pipes: no bubbly flow.

Air water, D = 25 mm, P = 1 bar.

APPLICATIONS : VertTD02

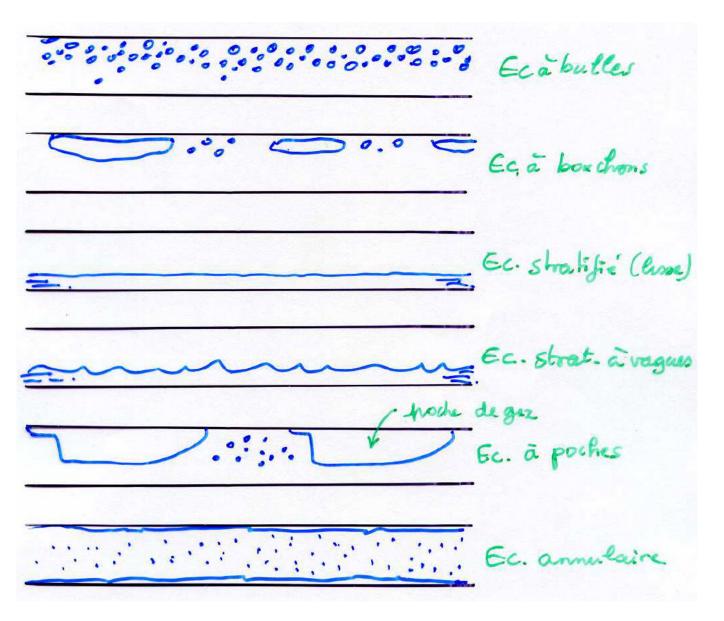


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FLOW PATTERNS IN HORIZONTAL FLOW



Main flow regimes:

- Bubbly
- Plug
- Stratified, smooth or wavy
- Slug of gas and plugs of liquid
- Annular

Modeling the transition based on mechanisms (Dukler & Taitel, 1986).

FLOW PATTERN IN HORIZONTAL FLOW



Photo 17, Wavy-stratified flow, with roll waves, Test 33

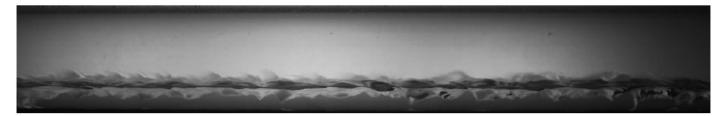


Photo 15, Wavy-stratified flow, Test 32.



Photo 25, Slug flow.



Photo 22, Annular flow.

Industrial occurrence and flow regimes

HORIZONTAL SLUG FLOW

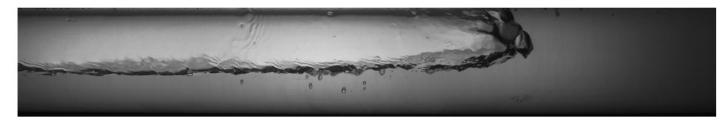


Photo 02.

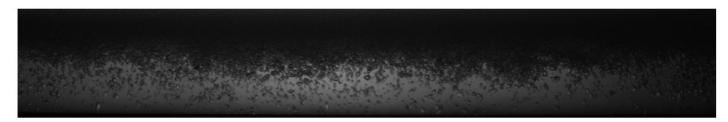


Photo 07.

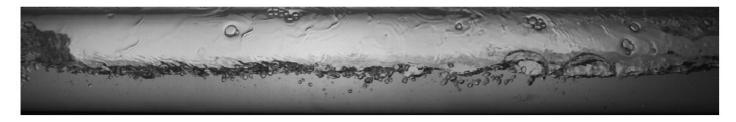
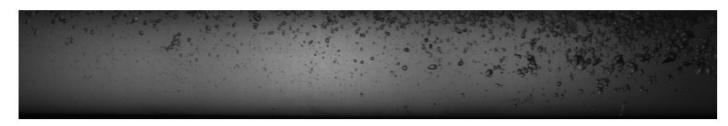


Photo 09.



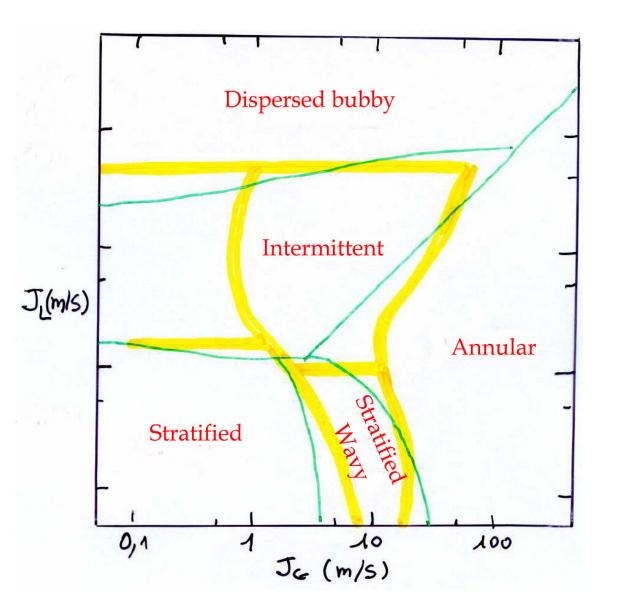


$$\alpha = 22\%$$

Industrial occurrence and flow regimes

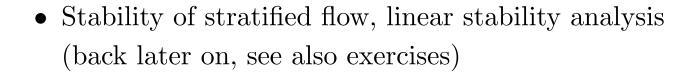


DUKLER AND TAITEL (1976) MODEL



Transition criteria:

- Empirical, e.g.: Mandhane, air-water, 25 mm, 1 bar.
- Mechanistic modeling of transition (Dukler & Taitel, 1986).

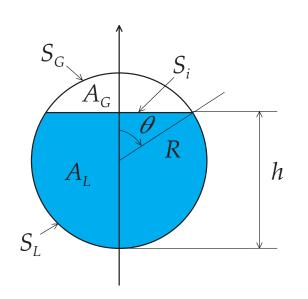


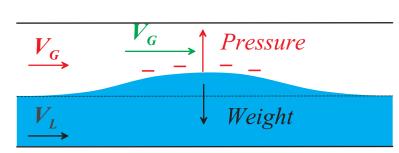
• The Kelvin-Helmholtz instability,

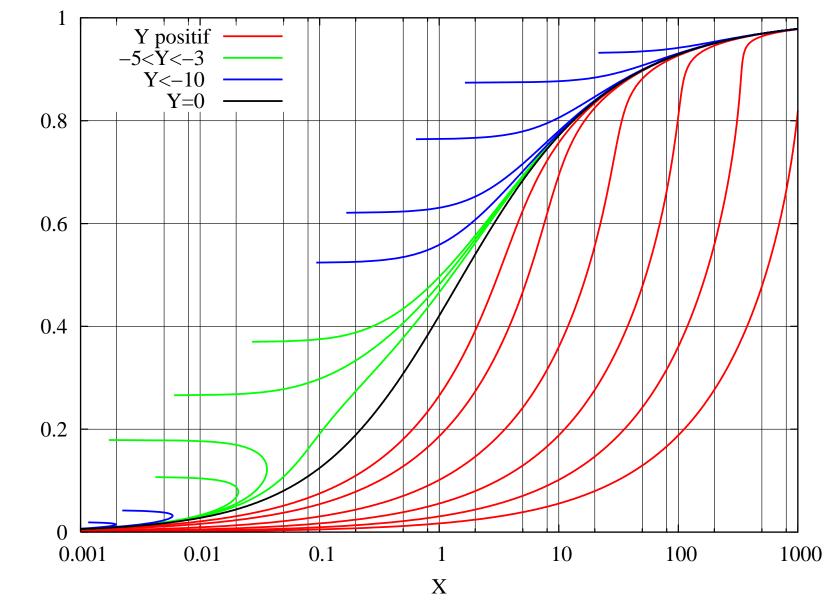
$$V_G \ge C_2 \left[\frac{(\rho_L - \rho_G) \cos \beta A_G}{\rho_G \frac{\mathrm{d}A_L}{\mathrm{d}h}} \right]^{\frac{1}{2}}, \ C_2 \approx 1 - \frac{h}{D}$$

• Base flow: smooth an horizontal interface

$$\begin{aligned} \tau_{G} \frac{S_{G}}{A_{G}} - \tau_{L} \frac{S_{L}}{A_{L}} + \tau_{i} \left(\frac{S_{i}}{A_{L}} + \frac{S_{i}}{A_{G}} \right) + (\rho_{L} - \rho_{G})g\sin\beta &= 0 \\ X^{2}f(A, D, A_{L}, P_{L}, D_{L}) - g(A, D, A_{G}, P_{G}, D_{G}, P_{i}) - 4Y \\ X^{2} &= \frac{\frac{1}{2}C_{L}\rho_{L}J_{L}^{2}Re_{LS}^{-n}}{\frac{1}{2}C_{G}\rho_{G}J_{G}^{2}Re_{GS}^{-n}} = \left| \frac{(\mathrm{d}P/\mathrm{d}z)_{LS}}{(\mathrm{d}P/\mathrm{d}z)_{GS}} \right|. \\ Y &= \frac{(\rho_{L} - \rho_{G})g\sin\beta}{\frac{4}{D}\frac{1}{2}\rho_{G}J_{G}^{2}C_{G}Re_{G}^{-m}} \end{aligned}$$







Industrial occurrence and flow regimes

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h/D

• Transition, stratified flow instability:

$$F^2\left(\frac{1}{C_2^2}\frac{\tilde{U}_G^2\frac{\mathrm{d}\tilde{A}_G}{\mathrm{d}\tilde{h}}}{\tilde{A}_G}\right) \ge 1, \quad \tilde{U}_G = \frac{A}{A_G}, \quad F = \left(\frac{\rho_G}{\rho_G - \rho_L}\right)^{\frac{1}{2}} \frac{J_G}{(Dg\cos\beta)\frac{1}{2}}$$

- Towards intermittent flow: $\frac{h}{D} \ge 0, 5$
- Towards annular flow: $\frac{h}{D} \leq 0, 5$
- Stratified smooth or wavy ?:

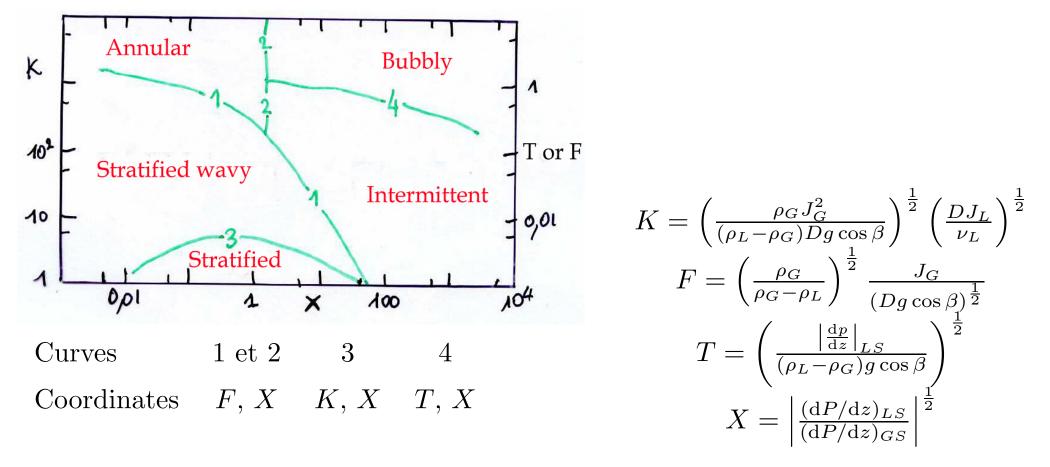
$$U_G \geqslant \left(\frac{4\nu_L(\rho_L - \rho_G)g\cos\beta}{s\rho_G U_L}\right)^{\frac{1}{2}}, \ K \geqslant \frac{2}{\tilde{U}_G\sqrt{s\tilde{U}_L}}, \ K^2 = \left(\frac{\rho_G J_G^2}{(\rho_L - \rho_G)Dg\cos\beta}\right) \left(\frac{DJ_L}{\nu_L}\right)$$

• Dispersed bubbles:

$$U_L \geqslant \left(\frac{4A_G}{S_i} \frac{g\cos\beta}{f_L} \left(1 - \frac{\rho_G}{\rho_L}\right)\right)^{\frac{1}{2}}, \ T^2 \geqslant \frac{8\tilde{A}_G}{\tilde{S}_i \tilde{U}_L^2 (\tilde{U}_L \tilde{D}_L)^{-n}}, \ T = \left(\frac{\left|\frac{\mathrm{d}p}{\mathrm{d}z}\right|_{LS}}{(\rho_L - \rho_G)g\cos\beta}\right)^{\frac{1}{2}}$$

Industrial occurrence and flow regimes

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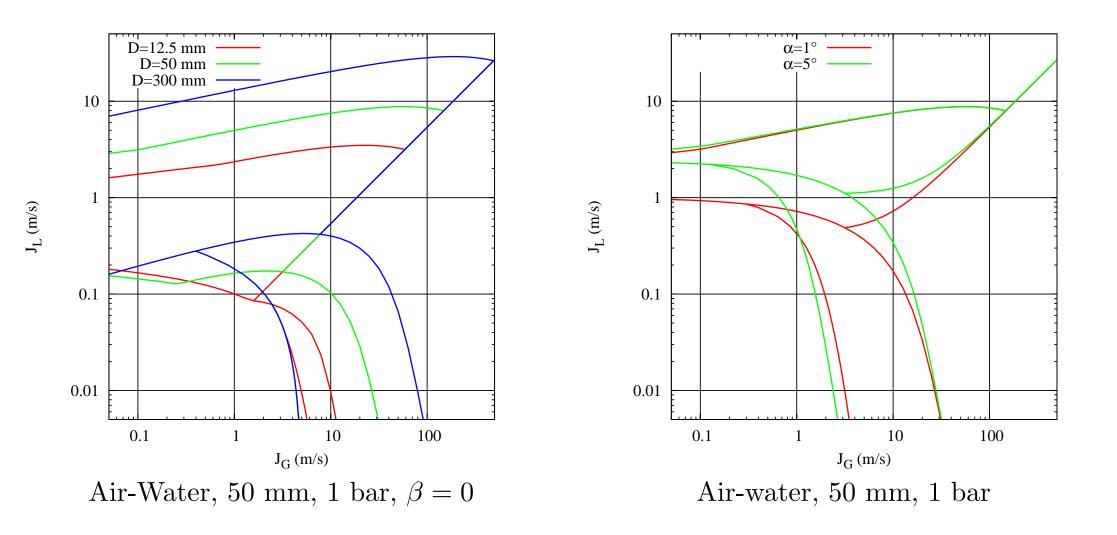


$$\left|\frac{\mathrm{d}p}{\mathrm{d}z}\right|_{S} = \frac{4C}{D}\mathrm{Re}^{-n}\frac{\rho J^{2}}{2}, \quad \mathrm{Re} = \frac{JD}{\nu}$$

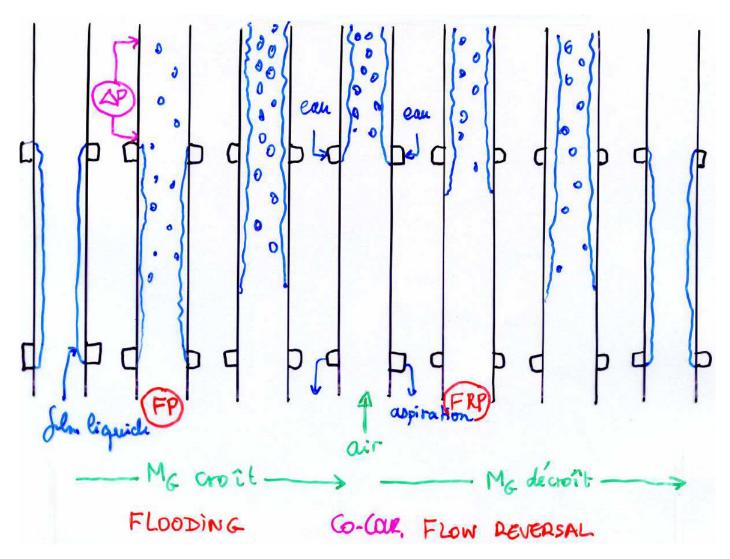
- Laminar: C = 16, n = 1. Turbulent: C = 0,046, n = 0, 2
- β : slope angle, $\beta = 0$, horizontal flow, $\beta > 0$, descending flows.

Industrial occurrence and flow regimes

APPLICATIONS : HoriTD03



FLOODING AND FLOW REVERSAL

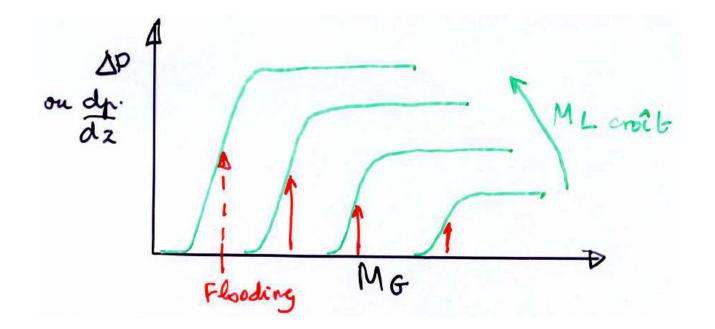


Flooding: transition from counter-current towards co-current up flow. Flow reversal: reverse transition

Industrial occurrence and flow regimes

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EXPERIMENTAL DETERMINATION OF FLOODING



Modeling of flooding and flow reversal see also Bankoff & Chun Lee (1986).



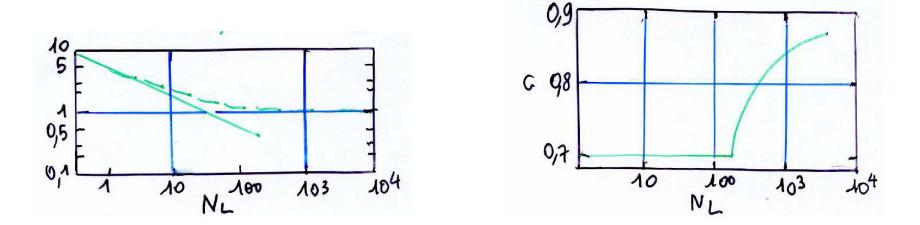
WALLIS MODEL

 $J_k^* \approx$ Froude number.

$$J_G^* = \frac{J_G \rho_G^{\frac{1}{2}}}{(gD(\rho_L - \rho_G))^{\frac{1}{2}}}, \quad J_L^* = \frac{J_L \rho_L^{\frac{1}{2}}}{(gD(\rho_L - \rho_G))^{\frac{1}{2}}}, \quad \boxed{J_G^{*\frac{1}{2}} + mJ_L^{*\frac{1}{2}} = C}$$

m and C depend on $N_L = \left(\frac{\rho_L gD^3(\rho_L - \rho_G)}{\mu_L^2}\right)^{\frac{1}{2}} \equiv Gr$

$$N_L > 1000 \begin{cases} m = 1 \\ 0,88 < C < 1 \text{ (smooth inlet)} , N_L < 1000 \\ C = 0,725 \text{ (sharp inlet)} \end{cases}, N_L < 1000 \begin{cases} m = 5,6N_L^{-1/2} \\ C = 0,725 \end{cases}$$



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FLOODING ET FLOW REVERSAL

- Wallis model: no pipe length effect.
 - Some experiments show J_G^* increase with the increase of L. Favors the wave instability mechanism.
 - Many specific correlations.
- Flow reversal
 - Wallis model, $J_G^*(FR) \neq J_G^*(Flooding)$, hysteresis effect, pipe diameter effect.

$$J_G^* = \frac{J_G \rho_G^{\frac{1}{2}}}{(gd(\rho_L - \rho_G))^{\frac{1}{2}}} = 0,5$$

- Puskina and Sorokin model,

$$Ku = \frac{J_G \rho_G^{\frac{1}{2}}}{(g\sigma(\rho_L - \rho_G))^{\frac{1}{4}}} = 3, 2$$

• Control mechanisms: still an open problem.

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- Delhaye, J.-M. 2008. *Thermohydraulique des réacteurs nucléaires*. Collection génie atomique. EDP Sciences.
- Dukler, A. E., & Taitel, Y. 1986. Multiphase Science and Technolgy. Vol. 2. Hemisphere. Chap. 1-Flow pattern transitions in gas-liquid systems: measurement and modelling, pages 1–94.

